

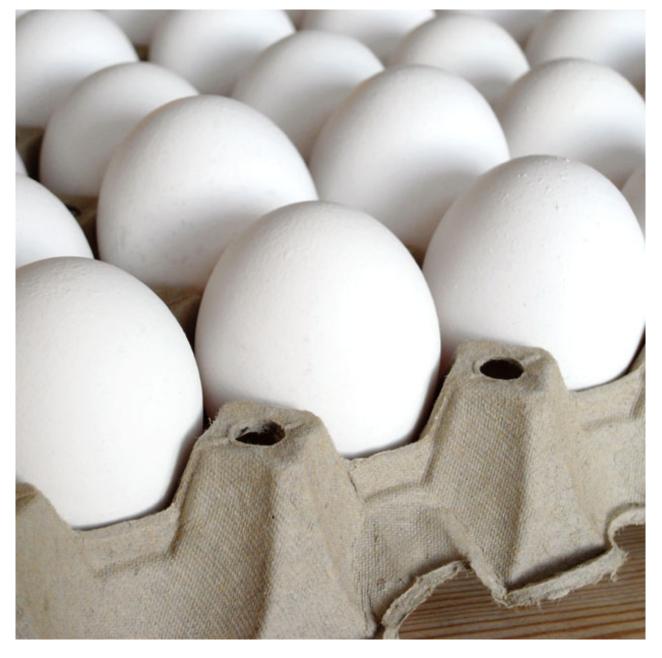




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## Towards the perfect soft boiled egg

48 Comments by Martin Lersch / April 9th, 2009



Many cookbooks suggest the following for boiling eggs: 3-6 min for a soft yolk, 6-8 min for a medium soft yolk and 8-10 min for a hard yolk. If you are satisfied with this, there is no need for you to continue reading. But **if you've ever wondered whether the size of an egg has any impact on the cooking time you should read on. And if you search the ultimate soft boiled egg we share a common goal!** From a scientific view point, a cooking time of approximately 3-8 minutes to obtain a soft yolk is not very precise. A number of important parameters remain unanswered: What size are the eggs? Are they taken from the fridge or are they room tempered? Are they put into cold or boiling water? And if using cold water – when should the timer be started? When the heat is turned on or when the water boils? And would the size of the pan, the amount of water and the power of the stove top matter?

#### A formula for boiling eggs?

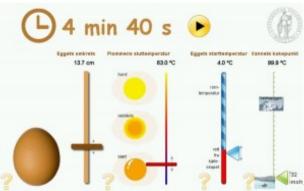
I still remember the very first time I heard about a formula to calculate the cooking time for eggs. I was in high school and as a recipe for the ultimate nerd the egg formula gave me a good laugh. Now – many years later – I count myself to this group of nerds  $\stackrel{...}{•}$  And thanks to the internet, google and Peter Barham's book  $\stackrel{...}{•}$  The Science of Cooking" – I have been able to find out much more. I haven't been able to track down the formula I heard mentioned, but the best documented formula nowadays is derived by  $\frac{Dr. Charles D. H. Williams}{*}$ , a lecturer in physics at University of Exeter. He has set up a nice page on the science of boiling eggs and there's even a pdf with the  $\frac{full derivation of the formula}{*}$ . Given the starting temperature of the egg  $T_{egg}$ , the temperature of the water

 $T_{\text{water}}$  and the desired temperature  $T_{\text{yolk}}$  (all in °C) at the yolk-white boundary, the cooking time t (in minutes) of an egg with mass M (in grams) is given by:

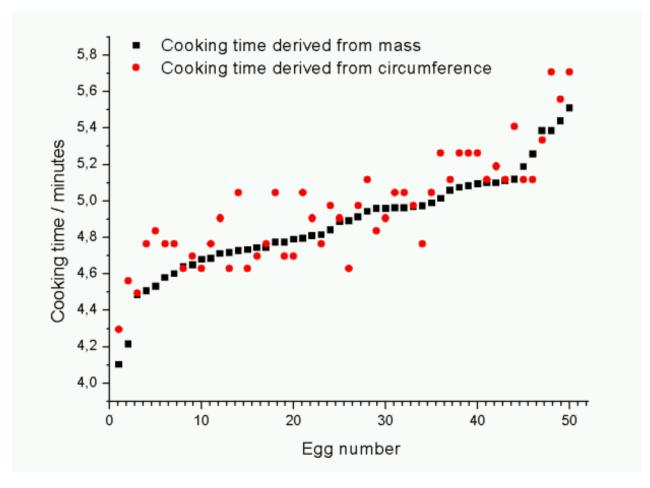
$$t = 0.451 M^{2/3} \ln \left[ 0.76 \times \frac{(T_{\text{egg}} - T_{\text{water}})}{(T_{\text{yolk}} - T_{\text{water}})} \right]$$

Whenever possible one should use weight measurements in the kitchen, but some times an accurate balance is not available and in those cases we can turn to the Peter Barham's formula which is published in <u>"The science of cooking"</u>. The circumference of an egg is easily measured around the thick end using a piece of string and a ruler. I used to have a **piece of string with three knots** at 13, 14 and 15 cm respectively to make it even simpler. The cooking time *t* (in minutes) for an egg with a circumference *c* (in centimeters) is given by:

$$t = 0.0152c^{2} \ln \left[ 2 \times \frac{(T_{\text{water}} - T_{\text{egg}})}{(T_{\text{water}} - T_{\text{yolk}})} \right]$$



Former colleagues of mine at the University of Oslo have made a nice <u>flash animation to do calculations with Barham's formula</u> if you're not too keen to dig out your calculator. Barham states that his formula gives the time for the *centre* of the yolk to reach the temperature  $T_{\text{yolk}}$  whereas Williams mention in the derivation of the formula that it calculates the time for the *yolk-white boundary* to reach  $T_{\text{yolk}}$ . I'm not able to tell whether the formulas actually differ in this respect or not (comments are welcome on this issue!). A comparison of the two formulas for a set of 50 eggs which I weighed and measured shows that for  $T_{\text{yolk}} = 63$  °C and  $T_{\text{water}} = 100$  °C they are quite similar, except for the larger spread of the circumference measurements (see plot below). For higher  $T_{\text{yolk}}$  or lower  $T_{\text{water}}$  Williams' formula consistently gives longer cooking times than Barham's formula. It remains to be seen which of the formulas will be closer to the truth.



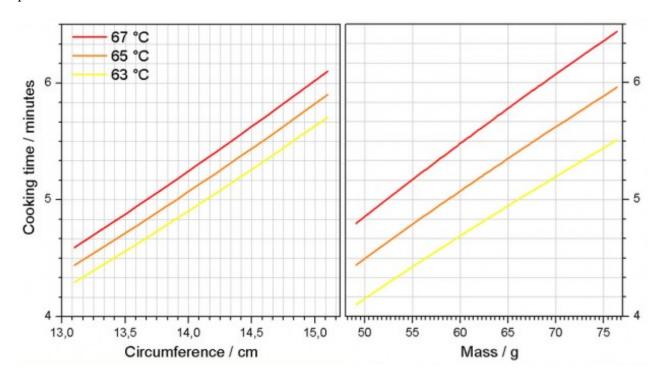
The graph shows the cooking time for 50 eggs (sorted by increasing mass) calculated from the mass and circumference using the two formulas shown above with  $T_{\rm yolk} = 63$  °C,  $T_{\rm water} = 100$  °C and  $T_{\rm egg} = 4$  °C. For the given conditions the two formulas give similar results. The most striking lesson learnt is that measuring the circumference is in fact not very accurate, hence the larger spread of these points.

The doneness of the egg depends on the temperature of the white and the yolk. Egg white starts to coagulate in the range 62-65 °C. At these temperatures it is the most heat sensitive protein, the ovotransferrin, which constitutes 12% of the egg white, which coagulates. The major protein of egg white, ovalbumin, makes up 54% of the white and doesn't coagulate until the temperature reaches 80 °C. The yolk begins to thicken around 65 °C and sets around 70 °C. Further heating to around 80-90 °C produces the crumbly texture typical of hard boiled eggs. Many of these changes are nicely illustrated in the picture of sous vide cooked eggs below, but the changes are also summed up in the following table:

Temperature / °C	Egg white	Egg yolk
62	Begins to set, runny	Liquid
64	Partly set, runny	Begins to set
66	Largely set, still runny	Soft solid
70	Tender solid	Soft solid, waxy
80	Firm	Firm
90	Rubbery solid	Crumbly texture

At sea level, the temperature of boiling water is 100 °C. At higher altitudes, the boiling is lowered. As a rule of thumb, the boiling temperature of water is lowered 0.3 °C for each additional 100 m above sea level. For an accurate calculation, check out his <u>calculator</u>. As we shall see later, the formula can of course also be used prepare eggs at sea level, using water kept at temperatures less than 100 °C. Lastly we must know the starting temperature of the egg which will typically be 4 or 20 °C.

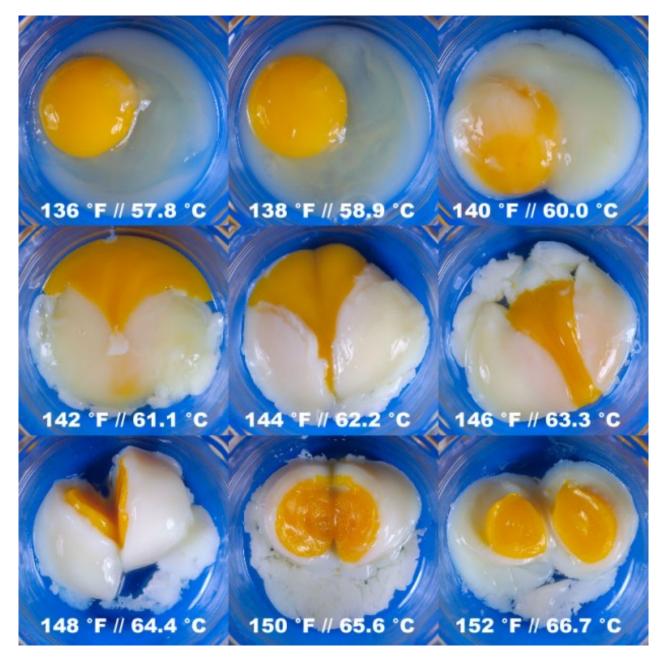
Based on  $T_{\rm water} = 100$  °C,  $T_{\rm egg} = 4$  °C and  $T_{\rm yolk} = 63\text{-}67$  °C I've prepared plots for the range of 50 eggs used in the previous graph. If the circumference or mass of an egg is known, the boiling time in minutes can easily be determined from the graphs. I've also prepared downloadable pdfs with the circumference and mass plots.



Cooking time for eggs with given circumference or mass to reach to reach 63, 65 and 67 °C respectively at the yolk-white boundary with  $T_{\rm water} = 100$  °C and  $T_{\rm egg} = 4$  °C (click for larger image or download pdfs with <u>circumference</u> and <u>mass</u> plots)

#### But is this the perfect egg?

No actually not... keep reading! The problem with using boiling water is that while you do heat the yolk to the desired temperature, you have virtually no control with the temperature of the white. If your water holds 95-100 °C, so will the white (or at least the outer most part of the white). This gives it a firm, rubbery texture. So the problem is, to put it differently, that we want to heat the yolk to somewhere above 65 °C, but we do not want to heat the white above 80 °C. The solution to this problem is to "boil" the egg at a temperature lower than 100 °C, which means not to boil it at all but rather sous vide it! Eggs are perfect for sous vide because you can just drop them into the water bath as they are. No plastic bags or vacuum packaging are required. Douglas Baldwin has cooked eggs sous vide for 75 min at different temperatures ranging from 57.8 to 66.7 °C as shown below. Notice how the egg whites and egg yolks change at the different temperatures.



Composite image of eggs cooked sous vide for 75 min at the indicated temperatures (Photo: <u>Douglas Baldwin</u>. Picture used with permission.)

The surprising thing with some of the sous vide eggs is that they are inverted (or opposite boiled). The white is still runny while the yolk is set. If you would like to try this but don't have a thermostated water bath for sous vide you can improvise a little. The thermostat most people do have in their kitchen is the baking oven (at least those with electric stoves). Preheat your oven to 70 °C. Then heat 1 L of water to 65-70 °C, put the eggs in, cover with a lid and leave the pan in the oven for one hour. The tricky thing here is that oven thermometers are notoriously wrong so use a separate handheld thermometer to check your oven. With some trial and error you should be able to obtain an inverted egg with a runny white and a yolk that has set.

Although scientifically amusing the inverted egg isn't really desirable form a culinary viewpoint — the white is a little to runny. Regrettably the formulas presented above aren't of much help either. They fail because they only take time and not temperature into account. **The perfect soft boiled egg in my opinion would have an egg white which is heated to around 70-80 °C and a yolk with temperatures ranging from 64 °C at the yolk-white boundary to about 60 °C in the center.** I guess it would be possible to prepare such eggs in a sous vide water bath held at 75-80 °C in less than an hour. A further complication of cooking eggs in real life is that they continue to cook when removed from the hot water. Normally this is alleviated by shocking the eggs in cold water, but if cooked at a lower temperature this could possibly be omitted. I will start experimenting to find a perfect mass-time-temperature combination with a time window that's as large as possible, and I'll report the results in a future blog

post. And these experiments will also include a test of the recipe for eggs cocotte by Joël Robuchon, found via Chubby Hubby's post on <u>slow-cooking an egg</u>.

#### **Exotic soft boiled eggs**



In his book "Off on a comet", science fiction author **Jules Verne shows** that he was actually aware of the possibility of "boiling" eggs at a temperature lower than 100 °C. He has correctly observed that water boils at lower temperature in high altitudes, and that on a fictional comet of appropriate mass, water will boil at 66 °C. The temperature is wisely chosen, because by keeping eggs at 66 °C, you really can't do anything wrong. From the last paragraph of the excerpt it seems that the eggs were not fully cooked after "a good quarter of an hour". Of course, there is also no mention about the size of the eggs, so any further speculations end here. But I'll rather leave it to you to read the excerpt from the <u>Gutenberg e-text version</u> – it's quite amusing:

The skillet was duly set upon the stove, and Ben Zoof was prepared to wait awhile for the water to boil. Taking up the eggs, he was surprised to notice that they hardly weighed more than they would if they had been mere shells; but he was still more surprised when he saw that before the water had been two minutes over the fire it was at full boil.

"By jingo!" he exclaimed, "a precious hot fire!"

Servadac reflected. "It cannot be that the fire is hotter," he said, "the peculiarity must be in the water." And taking down a centigrade thermometer, which hung upon the wall, he plunged it into the skillet. Instead of 100 degrees, the instrument registered only 66 degrees.

"Take my advice, Ben Zoof," he said; "leave your eggs in the saucepan a good quarter of an hour."

"Boil them hard! That will never do," objected the orderly.

"You will not find them hard, my good fellow. Trust me, we shall be able to dip our sippets into the volks easily enough."

The captain was quite right in his conjecture, that this new phenomenon was caused by a diminution in the pressure of the atmosphere. Water boiling at a temperature of 66 degrees was itself an evidence that the column of air above the earth's surface had become reduced by one-third of its altitude. The identical phenomenon would have occurred at the summit of a mountain 35,000 feet high; and had Servadac been in possession of a barometer, he would have immediately discovered the fact that only now for the first time,

as the result of experiment, revealed itself to him—a fact, moreover, which accounted for the compression of the blood-vessels which both he and Ben Zoof had experienced, as well as for the attenuation of their voices and their accelerated breathing. "And yet," he argued with himself, "if our encampment has been projected to so great an elevation, how is it that the sea remains at its proper level?"

Once again Hector Servadac, though capable of tracing consequences, felt himself totally at a loss to comprehend their cause; hence his agitation and bewilderment!

After their prolonged immersion in the boiling water, the eggs were found to be only just sufficiently cooked; the couscous was very much in the same condition; and Ben Zoof came to the conclusion that in future he must be careful to commence his culinary operations an hour earlier. He was rejoiced at last to help his master, who, in spite of his perplexed preoccupation, seemed to have a very fair appetite for breakfast.

There is in fact no need to head off to other planets to find examples of low temperature prepared eggs. If you go to Japan you'll find *onsen tamago* which litteraly translates to "hot spring eggs". Originally **baskets of eggs were lowered into hot springs**, but the temperature of hot springs vary so I imagine that there were several types of *onsen tamago* available (does anyone happen to know the exact temperature of the hot springs used?). After cooking the egg is typically cracked into a bowl of dashi soup with mirin and soy sauce. The challenge of preparing onsen tamago eggs at home is accurate temperature control (just as with sous vide in general). One tip I found was to place the egg on top of rice that has just cooked in a rice cooker. Leave the eggs to "cook" for about one hour while the "keep warm" function of the rice cooker is turned on.



Eggs boiled in <u>onsen</u> (japanese: hotspring), Nagano, Japan (<u>Photo</u>: Miya.m. Permission: GFDL, cc-by-sa-2.1-jp).

I've been told that in Finland some saunas are equipped with egg racks. Depending on where the rack is placed one could probably chose between hard boiled and soft boiled eggs. But the sauna would have to be kept warm for a long time due to the slow heat transfer from the hot air. And talking about eggs and saunas: If the eggs are placed directly on the hot stones they will not only be hard boiled, but actually turn completely brown and acquire a nutty flavor. In Korea such sauna eggs are known as Maekbanseok gyeran.

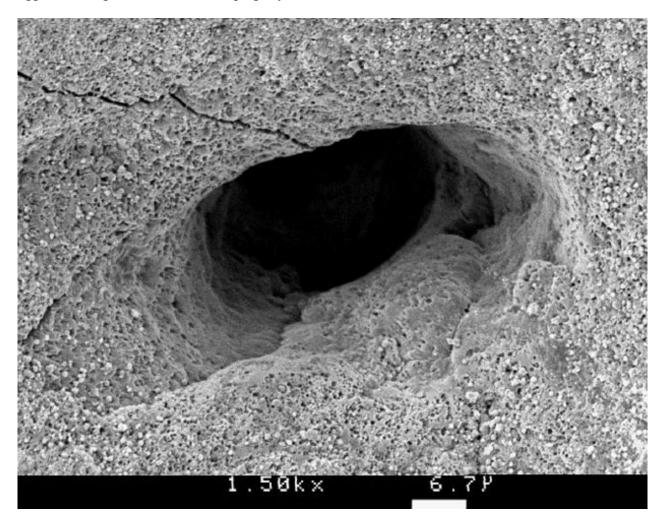
#### Other aspects to consider when boiling eggs

An egg has somewhere between 7000 and 17000 pores, meaning that water slowly evaporates (the density decreases from 1.086 g/cm³ by 0.0017 g/cm³ daily). This is also why eggs age faster at room temperature than in the fridge. Because of the pores, eggs should not be stored next to foods with a strong smell such as onions (unless of course, you want onion flavored eggs). **When boiling eggs it is not uncommon that they crack.** The most obvious reason is that they are dropped into the water and hit the bottom of the pot. Another reason for cracking is the expansion of trapped air at the blunt end of the egg. This air cannot escape fast enough through the small pores. Conventional wisdom has it that piercing a small hole in the blunt end will let expanding air escape to avoid cracking. It turns out <u>someone</u> has actually scientifically tested this (with 1000 eggs) and their finding was that there was little cracking for fresh eggs, regardless if they were pierced or not. Piercing reduced the cracking of 5-day old

eggs and totally eliminated cracking of 28-day old eggs. The authors theorize that the air pocket grows due to evaporation (meaning there is more air to expand) and that the egg shell of fresh eggs is porous but that the pores gradually become clogged upon storage. Curiously the <u>abstract</u> concludes with the following sentence (this was written in 1973, but it's still quite unusual for a scientific journal):

Housewives should pierce eggs before boiling them, since if they are fresh it will do no harm and if they are stale it will prevent splitting.

We can safely assume that the advise holds true for men as well! Apart from piercing holes to avoid cracking it is possible to reduce the potential damage from cracking by addition of salt or vinegar to the water. This will help the egg white coagulate faster and thus plug any crack formed.



Picture of egg shell pore (Photo: <u>Jim Ekstrom</u>. Permission: Freeware for non-commercial use).

If you've read this far, make sure to also read <u>how the egg yolk problem was finally solved</u> and my <u>follow up post with pictures and a video</u> of egg yolk cooked at 63.0 °C for 40 to 155 minutes!

20 Like {1.2k

Tags: boiling eggs, Charles Williams, chemistry, cocotte, cooking, easter, egg, eggs, everyday science, formula, hard boiled, Jules Verne, maekbanseok gyeran, molecular gastronomy, onsen tamago, perfect eggs, Peter Barham, popular science, Robuchon, sauna, soft boiled, sous vide

Filed under: experiments, molecular gastronomy

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« Previous: TGRWT #17: apple and rose

Next: Cooking by ratios - new book by Ruhlman »

#### **Comments**

1. Martin Lersch Says:
April 9th, 2009 at 12:59 am

There were a lot of temperatures in this post so I found a widget for Celcius-Fahrenheit conversions. I placed it at the top of the right sidebar – hope it works.

2. <u>Leela@shesimmers</u> Says:

April 9th, 2009 at 6:10 pm

This has got to be one of the greatest food blog posts I have ever come across. Thank you so much for all the hard work and research that have gone into this.

I guess the fact that I find this to be fastinating means I'm also a nerd. But, hey ... 😃

- 3. <u>» Another good one from Khymos perfect boiled eggs Bringing food chemistry to life</u> Says: April 9th, 2009 at 8:54 pm
  - [...] http://blog.khymos.org/2009/04/09/towards-the-perfect-soft-boiled-egg/ [...]
- 4. Amanda from Mrs. W's Kitchen Says: April 10th, 2009 at 1:02 pm

Very, very interesting! I find boiling eggs such an elusive goal—whatever stage you want it to reach.

Of course I've been cheating-steaming them in an egg-a-muffin machine. I know. I'm such a faker.

5. Peter F. Says:

April 13th, 2009 at 4:49 pm

The upper temperature at onsens and other hot springs for bathing are  $42^{\circ}\text{C} - 49^{\circ}\text{C}$  ( $108^{\circ}\text{F}-120^{\circ}\text{F}$ ), at most. Any hotter and scalding would occur. This seems to be too low to effectively cook a good soft boiled egg.

Though natural hot springs may bubble from the ground at any temperature, so it is possible the pools used to boil the eggs are much higher temperature.

6. Martin Lersch Says:

April 13th, 2009 at 5:41 pm

Could it be that the eggs are cooked in hotter parts of the onsens that are not used for bathing? At the temperature you mention nothing will happen to the egg.

7. Peter F. Says:

April 13th, 2009 at 6:56 pm

Asked a Japanese co-worker about it. You are correct in that the onsens used to make onsen tamago are not used for bathing, they are much hotter ideally around 70° give or take according to him.

He made mention of some japanese chefs that at one point experimented with cooking sous vide in the onsens, not much more info than that interesting snippet

8. *Hiroki* Says:

April 14th, 2009 at 3:13 am

In Goshogake onsen, the temperature is  $90^\circ$ . An egg becomes black. In Ohiradai onsen which is part of Hakone,  $65^\circ$ . It takes 30 min to make onsen tamago.

## 9. *Martin Lersch* Says: April 14th, 2009 at 7:36 pm

Thanks for the info on the onsen temperatures. I can imagine that the 65-70 °C you mention should be ideal.

I've read about the black onsen eggs (kuro tamago) – quite interesting. The black color is caused by hydrogen sulfide which reacts with iron from the hot spring to form iron sulfide which is deposited on the egg shell. This reaction does not occur in the hot spring due the low pH, but occurs on the egg shell which is primarily calcium carbonate. When left in the air the iron (II) sulfide is oxidized by oxygen in the air to iron (III) oxide. The sulfur ends up as either elemental sulfur or sulfur dioxide.

### 10. Olly Rouse: Sous Vide Solutions Says: April 15th, 2009 at 10:42 am

You may find this an interesting one to play with....

A client wanted to use a dual chamber Clifton waterbath which he wanted to cook perfect soft boiled eggs in, by unskilled operators, and to have them ready is seconds, for a period lasting breakfast service [2hrs]...

This is what I worked out for him:

- 1. In a filled dual chamber water bath, set the left hand side to 90°C and the right hand side to 55°C [place a lid on the left side]
- 2. Once the bath is up to temperature, place [66g free range eggs] into the waterbath set at 90°C. These eggs need to be cooked from room temperature [20°C] and re-cover with the lid
- 3. Immediately start a timer for [7:30mins]
- 4. Once the timer has gone off, immediately lift the eggs from the left side chamber, and place into the right hand side, set at  $55^{\circ}$ C
- 5. Allow eggs to remain in the right side bath for a minimum of 5 minutes.
- 6. The eggs are now ready to serve 'Soft Boiled'
- 7. The eggs will remain the same upto 4 hours after placing into the 55°C bath.

Have fun trying it out!

Olly Rouse

### 11. *Martin Lersch* Says:

April 15th, 2009 at 10:28 pm

Olly: That's an interesting approach for those with access to a dual chamber sous vide! I've tried to insert the numbers you mention (66 g egg, 90 °C water, eggs starting at room temperature and 7:30 min cooking time) into the formulas using the linked flash animation. This suggests to me that the yolk would end up well above 70 °C. Is there something I missed or does this only show that the formulas are less reliable at lower cooking temperatures? Or does the efficient cooling at 55 °C play an important role here?

### 12. Olly Rouse: Sous Vide Solutions Says:

April 16th, 2009 at 12:56 am

Martin...

Yea, the drop in temperature to 55°C is the important thing here.

The eggs I used came out as perfect soft boiled eggs to eat immediately after cooking at 90°C for 8 minutes. I mean by 'perfect' that the white is hardened, the shell cracks well, and the yolk bright and runny: A soft-

boiled egg to eat out of the shell for breakfast with toast soldiers [a very popular breakfast in the UK]

I then needed to work out the latent heat the shell and white had, which would carry on cooking the yolk, hardening it as the temperature reached its coagulation point [68°C]

Ran a series of tests and got the result fairly quickly.

Opening the egg after the 7:30 minutes at 90°C resulted in a still runny egg white.

Opening the egg after 8:00 minutes at 90°C, and then holding at 55°C for 10 minutes resulted in a thick, semi-solid yolk.

Hope this makes it a little clearer, and that you get to try it for yourself....

You could always accurately set the cooking chamber to 90°C, cook the eggs for 7:30 minutes, and then top up the cooking chamber with cold water to bring down the water temperature rapidly – that should work fine  $\Box$ 

Also, we did trials on cheap battery farmed hens eggs, and these needed upto 30 seconds less cooking time at the higher temperature.

#### 13. robert millman Says:

April 17th, 2009 at 1:54 pm

Other things that I have found that add to variations in sous vide cooking that are hard to account for are the type of chicken, type of feed and season. All of these contribute greatly to the protein and water profile of an egg which I have found can lead to almost a degree difference in reaching the same consistancy with different sourced eggs.

#### 14. johnno Says:

April 17th, 2009 at 11:30 pm

I find if I sous-vide eggs at 64.5C for an hour and them crack them open, the yolk is perfect but the egg white is sometimes less set than desired. Very easy resolution, a blast with the propane torch fixes this quickly and the wonderful custard egg yoke stays constant insulated by the now firmer egg white. Also Martin, the immersion circulators increment by 0.1 degress, (well the polyscience one does) why was your test increment 1.1 degree?

#### 15. *Martin Lersch* Says:

April 19th, 2009 at 8:34 pm

Olly: Thank's for all the details regarding your sous-vide eggs! This suggests that cooking of eggs takes a little longer than the formulas suggest.

Robert: Interesting! I'm wondering – would a higher protein content make the egg more or less heat tolerant?

johnno: Or you could just quickly dipp the egg into boiling (or near) boiling water:). Regarding the pictured sous vide eggs – this experiment was in fact done by Douglas Baldwin and he did the experiment in Fahrenheit.

#### 16. Blil T. Says:

April 21st, 2009 at 1:52 am

You write, "I will start experimenting to find a perfect mass-time-temperature combination with a time window that's as large as possible...." Instead of the expensive sous-vide approach, would it be possible to instead do "eggs in cooling water", akin to fish in cooling water? Then all the poor home gastronomist needs is a scale and a timer.

Thanks, Bill

#### 17. Rachel Says:

May 18th, 2009 at 5:16 pm

THis was a very interesting post and made me want to run to the kitchen to start experimenting! Only problem: I'm allergic to eggs and it seems a waste to cook something you can't eat!

### 18. <u>Budget sous-vide cooking | Marijn.nu</u> Says: June 14th, 2009 at 8:45 pm

[...] while the egg white stays soft. For more information about the perfect boiled egg, I recommend this [...]

#### 19. donald Says:

June 23rd, 2009 at 2:45 pm

This is not really off-topic, but does not address perfection of softboiled eggs. Here is the story:

Today we had soft-boiled eggs for breakfast. As always, we pricked the ends of the shell with a needle so as to reduce cracking. One of the eggs tended to float when placed in the boiling water—not a good thing. This egg immediately leaked through the puncture, and over a minute or so ejected most of its contents. The yolk formed a perfect ball in the boiling water. It looked exactly like a yolk removed from inside a hardboiled egg. The egg shell, meanwhile, remained intact. The contents were ejected through the original pinhole and cracks in the shell (post-mortem, it appears that the egg was cracked to begin with and may have had trapped gasses from incipient decomposition). I am baffled by the mechanism whereby the yolk remained intact in the boiling turbulent water after being squirted through a small aperture. Any ideas?

### 20. <u>Hvordan koke perfekte egg</u> Says: July 15th, 2009 at 10:56 am

[...] motsatt side av skalaen kan man legge ut på en titimers ferd mot et perfekt bløtkokt egg, slik som i denne imponerende posten fra Khymos.org. Både prosjektet og nettstedet hans er lesverdig, men det blir likevel litt mye. Lærdommen derimot, [...]

#### 21. Jason Says:

July 22nd, 2009 at 12:41 am

it sounds like you've done a lot of research into boiling eggs. being an egg lover myself, i respect that a lot. i lived in japan for 6 months for work and while i was there ate at the little ramen restaurants around tokyo all the time. Almost all of them allow you to order an egg alongside your ramen which is brought out with the shell intact and prepared in such a way that you can crack it like it's raw with the egg sliding out like it's softboiled. no peeling involved whatsoever. if you could check out my video of the egg here: <a href="http://www.youtube.com/user/jasonyeh#play/uploads/4/5R4uuUO6Q3A">http://www.youtube.com/user/jasonyeh#play/uploads/4/5R4uuUO6Q3A</a>

and then either comment with your explanation on how that's done or email me back, i'd be most appreciative.

cheers.

#### 22. Olly Rouse: Sous Vide Solutions Says:

July 22nd, 2009 at 3:16 pm

Jason

What you describe and the video link is the 'slow cooked egg' as detailed by Douglas Baldwin above.

Many restaurants have been serving this for a few years now as vegetarian dishes etc.

Hope this allows you to do it yourself!

Have fun

Olly

### 23. *The science of eating* Says:

July 28th, 2009 at 11:11 am

[...] a simple thing like a soft-boiled egg. So many options, so many [...]

#### 24. David Harmon Says:

July 28th, 2009 at 11:06 pm

I bow before your gastronomic geekery! My usual way of cooking eggs is simply to use one of those plasticegg thermosensors (they change color from red to purple as the egg cooks), rather than timing the pot. I go for hard-boiled rather than soft, but this gizmo lets me do nice hard-boiled eggs without overcooking them, and further lets me start the batch with the eggs in cold water without worrying about corrections.

Regarding cracking, my "oral" tradition is to add salt and vinegar to the boiling water — purportedly, the salt helps quickly coagulate leaks, while the vinegar makes the eggs easier to peel afterwards. Perhaps you have more information about these additions?

#### 25. Kemmy Says:

August 3rd, 2009 at 8:39 am

OMG I LOVE EGGS !!!!!!!!!!!!!!!!

#### 26. algee Says:

August 31st, 2009 at 9:37 pm

easiest way to peel the egg i've found is to plunge the cooked egg (hard-boiled, as that's the only way i like em) in ice water until it's cold. crack the shell all over, then start peeling (i usually start at the big end where the air hole would be). the egg contracts and the shell pulls right off with minimal damage to the white.

ps. great blog.

#### 27. Slow Cooker Guy Says:

September 8th, 2009 at 6:48 am

Very interesting. I'm definitely having trouble in the perfect soft boiled egg department.

Armed with new info, it's back to the kitchen for me. Might try this again with some ramen.

#### 28. Michael Long Says:

September 9th, 2009 at 5:53 am

OMG... I love eggs and always trying to prefect the perfect egg. I'm played with slow-cooking eggs in water baths. And any other way I can find. And it's a hit or miss thing. But I'll get there...

#### 29. Kim (Edible/Usable) Says:

September 17th, 2009 at 6:32 pm

Great blog and great post about eggs. I wrote a post about Egg Graders (<a href="http://www.edibleusable.com/2009/03/useful-things-ill-never-use-acme-egg-grader.html">http://www.edibleusable.com/2009/03/useful-things-ill-never-use-acme-egg-grader.html</a>) awhile ago and learned a lot about eggs that I never knew. I wonder how actual egg grade (freshness) affects the time they take to cook as well... I've taken it for fact that at least for poaching eggs, it's best to use fresh, room

temperature eggs – but I think that has more to do with the egg white holding together in the water.

#### 30. William Loughborough Says:

November 7th, 2009 at 4:27 pm

No mention of what were called "dropped eggs" in the MIT kitchen where the cooks were Jewish and the eggs were dropped into a large swirling vat of nearly boiling water and lifted out with slotted spoons as the whites coagulated. The yolks were appropriately runny and each egg stayed fairly intact. This was in 1942.

When I try it at home with fairly large saucepan, it "almost" works, but I feel like it's quite close, still need some training on scooping just right.

Love.

# 31. <u>Paul Delahunt-Rimmer</u> Says: December 24th, 2009 at 9:59 am

This is a fascinating article. My wife can boil an egg to perfection I cannot. When we were serving eggs to order on Royal and Ministerial flights it was always a problem for the stewards to judge the exact time for a soft boiled egg with a cabin altitude of about 8,000ft. Their solution was a little easier than the mathematics displayed on this site. They put a pile of eggs into the boiling water (<100°C of course). Then at intervals they kept taking one out and opening it until they got the correct result. The rest were then ready. I'm sure our scrambled eggs on the flight deck were made of the first underdone boiled ones but we didn't mind.

#### 32. David D White Says:

December 29th, 2009 at 5:51 pm

Just curious about the article. Thanks for posting.

#### 33. *seth* Says:

January 16th, 2010 at 1:46 am

yea yall need to put some research articals on here

#### 34. Alexia Parada Says:

January 28th, 2010 at 8:29 am

Hard to understand..

#### 35. *George* Says:

February 5th, 2010 at 1:01 pm

I thought it would be interesting to start researching on the subject. I would buy thermometers and lots of eggs. But I found this.

Great article.

But now I have to find something else to research.. :/

#### 36. » KOKEILLEN #1 – 68C:n kananmuna Says:

March 20th, 2010 at 7:27 pm

[...] Aiheesta voi lukea hienoja pohdintoja ja systemaattisesti toteutettuja kokeita esimerkiksi täältä tai täältä [...]

### 37. <u>Gigabiting » Blog Archive » Just in Time for Easter: learn how to boil an egg</u> Says: March 23rd, 2010 at 7:23 pm

[...] the food science experts at Khymos to calculate precise cooking times. You'll find formulas that account for factors like egg [...]

#### 38. *Par olām* Says:

March 31st, 2010 at 3:08 pm

[...] gastronomijas entuziastiem olu vārīšana ir vesela māksla. Savukārt Japānā olas vāra karstajos avotos un Korejā īsta delikatese esot [...]

# 39. *Ik. de man van ...- Pinkdot* Says: April 5th, 2010 at 9:22 pm

[...] dit op een eigen wijze willen bereiken. Prikken we wel of geen gaatje in het ei! Zij niet, ik wel. Wetenschappelijk kan ik dit onderbouwen maar daar gaat het niet om bij een micro-conflict. Het gaat om de krachtmeting, het [...]

# 40. <u>Home Cooked « Food In Singapore and the Rest of the World</u> Says: April 23rd, 2010 at 3:55 am

[...] I have a soft spot for Sous vide. I do believe that it is the future of home cooking, once the cost of units such as the Sous vide Supreme starts to decrease. I bought a Sous vide magic device and hooked it up to my rice cooker as part of my plan to discover what Sous vide is capable of, first hand. And this was the first thing I cooked- The Perfect Egg [...]

# 41. מה קורה כשמבשלים ביצים? | מדע בצלחת Says: July 29th, 2010 at 7:58 am

[...] ביצים נראה פשוט להפליא, אולם מסתבר שזו סוגיה מדעית מורכבת וקיימות כלפיה גישות שונות, כל אחת מהן בעלת [...] הסבר [...]

# 42. <u>Der Weg zum perfekt weich gekochten Ei | Sofa-Kurier</u> Says: August 8th, 2010 at 9:24 am

[...] und zwar dort im blog fand ich die folgende Anleitung für das perfekte weich gekochte Ei. Towards the perfect soft boiled egg Der ganze Artikel wäre zu lang, um ihn hier einzufügenDie Seite khymos wird betrieben von [...]

### 43. <u>מה קורה כשמבשלים ביצים? | סלונה</u> Says: August 8th, 2010 at 5:23 pm

ביצים נראה פשוט להפליא, אולם מסתבר שזו סוגיה מדעית מורכבת וקיימות כלפיה גישות שונות, כל אחת מהן בעלת [...] הסבר [...]

#### 44. John Says:

August 9th, 2010 at 2:22 pm

"No mention of what were called "dropped eggs" in the MIT kitchen where the cooks were Jewish and the eggs were dropped into a large swirling vat of nearly boiling water and lifted out with slotted spoons as the whites coagulated. The yolks were appropriately runny and each egg stayed fairly intact. This was in 1942.

When I try it at home with fairly large saucepan, it "almost" works, but I feel like it's quite close, still need some training on scooping just right. "

This is the way Restaurants "do" Poached eggs in large quantities. Did you Acidulate the water – usually Lemon juice.

- 45. <u>Egg Safety: How to Boil an Egg | Gigabiting</u> Says: August 26th, 2010 at 1:28 pm
  - [...] the food science experts at Khymos to calculate precise cooking times. You'll find formulas that account for factors like egg [...]
- 46. בישול ביצים: מסביב לעולם בשמונים דרכים | מדע בצלחת Says: September 19th, 2010 at 9:19 pm
  - [...] לבישול אטי של ביצים. הביצים מוכנסות למי המעיין בסלסילה ומבושלות בקליפתן בתוך המים החמים כ-30 [...]
- 47. Doug Says:

September 24th, 2010 at 10:44 pm

With the yolk firming up before the white, I would suggest a two part cooking method (something I will try this weekend). Sous Vide the egg at 64.5 degrees for some period of time to let the entire egg cook through. This will produce a perfect yolk and an under cooked white. Remove the egg and refrigerate overnight.

Then, the next day, take the refrigerated egg and place it in a sous vide of something slightly over 80 degrees (testing required here as my guess is around 82 or 84.) Leave it in the sous vide only long enough for the yolk/white boundary to reach 64.5. Remove, shock the shell with ice water for a moment to stop it from heating the egg further. Wait a minute or two for all the residual heat in the egg to stabilize and serve.

This will allow the second heating to cook the white while using the refrigeration to insulate the perfectly cooked yolk from overcooking.

Anyone with any thoughts or comments?

48. <u>The 64-degree poached eggs | Kayahara</u> Says: September 27th, 2010 at 6:58 pm

[...] to. You can see a good overview of the effects of different temperatures at Martin Lersch's Khymos blog, and in the Cooking Issues [...]



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Martin Lersch 22 Mar

@tastymolecules

That's the intention of CC bync-sa:) RT @sciencegeist: ... today I'm borrowing heavily from @tastymolecules Texture

blog.khymos.org/recipe-colle

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